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1 Questions: A) Determine the molecular formula for the
following hydrocarbons: a. Hexane = C_6H_{14} b. Propyne = C_3H_4 c. 2-octyne = C_8H_{14} d. Decane = $C_{10}H_{22}$ e.
4-nonene = C_9H_{18} B) Describe the difference between a
saturated hydrocarbon and an unsaturated hydrocarbon.
Name a total of 3 saturated hydrocarbons and a total of 3
unsaturated hydrocarbons from Data Table 1 and Data Table
2.

ANSWERS

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